

Review



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The impact of air pollution on terrestrial managed and natural vegetation

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Although awareness that air pollution can damage vegetation dates back at least to the 1600s, the processes and mechanisms of damage were not rigorously studied until the late twentieth century. In the UK following the Industrial Revolution, urban air quality became very poor, with highly phytotoxic SO₂ and NO₂ concentrations, and remained that way until the mid-twentieth century. Since then both air quality, and our understanding of pollutants and their impacts, have greatly improved. Air pollutants remain a threat to natural and managed ecosystems. Air pollution imparts impacts through four major threats to vegetation are discussed through in a series of case studies. Gas-phase effects by the primary emissions of SO₂ and NO₂ are discussed in the context of impacts on lichens in urban areas. The effects of wet and dry deposited acidity from sulfur and nitrogen compounds are considered with a particular focus on forest decline. Ecosystem

eutrophication by nitrogen deposition focuses on heathland decline in the Netherlands, and ground-level ozone at phytotoxic concentrations is discussed by considering impacts on semi-natural vegetation. We find that, although air is getting cleaner, there is much room for additional improvement, especially for the effects of eutrophication on managed and natural ecosystems.

This article is part of a discussion meeting issue 'Air quality, past present and future'.

1. Introduction

There is a very long history of understanding that air pollution is a threat to vegetation. The medieval writer and mystic Hildegard von Bingen (1098–1179) noted in her book *Causae et Curae* [1] that dust within rain was believed to damage crops. In the 1600s developments in early scientific understanding did not miss the importance of pollutants from combustion processes and industry in damaging plants, including the first evidence of air pollution as a transnational concern. In the first decade of the 1600s, King James passed an act 'against burning of Ling, and Heath, and other Moor-burning . . .' noting that 'some parts even of France itself lying South west of England, did formerly make of being infested with Smoakes driven from our Maritime Coasts, which injur'd their Vines in Flower' [2].

However, prior to the nineteenth century and the Industrial Revolution, there were no air quality measurement networks, and evidence that air pollution was damaging vegetation was limited to areas in the immediate vicinity of point sources (e.g. lime kilns, charcoal production, early smelting activities). In *Sylva* [3], John Evelyn recognized the threat to vegetation through being 'infected with fogs and poys'nous vapours, or expos'd to sulfurous exhalations', while Fabri [4] saw volcanic emanations as a source of acidic damage to fruit. As industrialization increased, so did reports of environmental damage: in the summer of 1794, a visitor to south Wales observed, 'Nearer to Swansea, there are extensive works both of copper and iron, from the malignant influence of which every trace of vegetation in the neighbourhood has from a very short period after their erection, been totally annihilated' [5]. At this time the main phytotoxic pollutant was thought to be SO₂ and to a lesser extent HCl and NO₂.

An analysis of global sources of the major gaseous pollutants by Hoesly *et al.* [6] showed the dominance of sources of SO₂ and NO_x in Europe and North America from 1880 until 1980 (figure 1). Thereafter, following the introduction of control measures in Europe and North America and the rapid growth of sources in Asia the main sources of SO₂ and NO_x have been in East and South Asia (figure 1). Early records of pollution impacts included observations from both the natural and agricultural environments. From around 1800 the dark or melanic form of some species of moths in England began to increase in frequency. On soot-covered tree trunks, darker moths were better camouflaged and thus more avoided predation than light ones [7]. Sheep fleeces became blackened by smoke [8] and in the industrializing US bird feathers became covered in black carbon particles [9]. 'Black rain' was observed in remote areas of Scotland [10] and soot from Manchester's factories caused black snow in Scandinavia as described by the playwright Ibsen in his play *Brand* in 1867. Interestingly, the latter observation shows clear evidence of long-range transport of pollutants in Europe long before the arguments of the 1970s on whether industrial emissions from the UK, Germany and France contributed to acid deposition in Scandinavia.

The rapid increase in both industrial and domestic SO₂ and NO₂ emissions in the early twentieth century in several European countries, notably the UK, Germany and France, and in North America (figure 1), generated regional surface concentrations sufficient to damage both crop- and wild plant species as demonstrated by the following examples. An extensive series of field investigations by Cohen and Ruston published in 1925 [11] substantiated the observations of various earlier workers on the effects of the coal-smoke polluted atmosphere on city vegetation. Using a transect from the suburbs towards the centre of Leeds, Cohen and Ruston [11] found

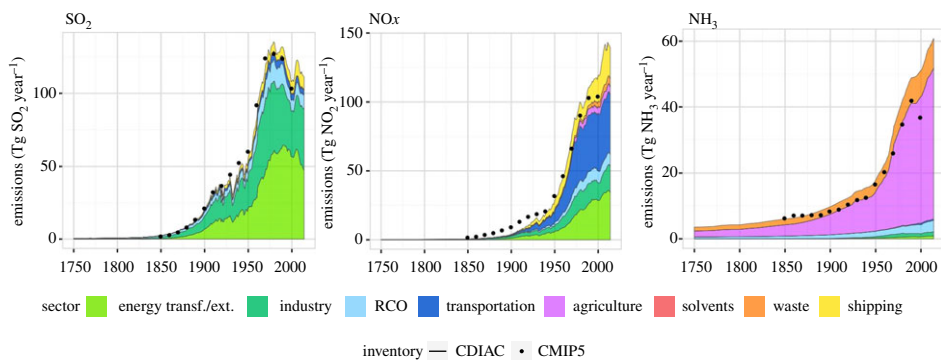


Figure 1. Global emissions of SO_2 , NO_x , and NH_3 between 1750 and 2010. Adapted from Hoesly *et al.* [6]. (Online version in colour.)

depressions in yield, various degrees of visible damage and alterations in growth habit of a number of species (e.g. *Laurus nobilis*), which increased in intensity as pollution levels increased towards the city centre. The ill effects on vegetation were positively correlated with the annual wet deposition of particulate matter and sulfur compounds, but the relative toxicity of the different components of the coal smoke was not determined. In 1928, Pettigrew [12] described extensive damage to plants and the total failure of ornamental trees and shrubs in Manchester parks as a result of atmospheric pollution, while in 1941 Metcalfe [13] described the pollution-induced premature shedding of leaves, flowers and buds on ornamental species (e.g. *Begonia foliosa*) in towns during periods of foggy weather.

Much of the earlier interest in the effects of air pollution on vegetation in the UK focused on such damage to ornamental species in urban areas, where losses were considered to be more serious from an aesthetic rather than economic viewpoint (e.g. [11–13]). However, in 1952, Bleasdale [14] devised an experiment to show that a coal-smoke polluted atmosphere can considerably depress the yield of the economically important strain of the pasture grass *Lolium perenne*. When attempts were made to improve the quality of hill pastures in heavily polluted districts of East Lancashire by reseeded with this strain of *L. perenne*, the plants became established successfully, but soon declined in productivity and eventually disappeared. Bleasdale grew the plants in Manchester in two greenhouses: one ventilated with polluted ambient air and the other with air purified by passage through a water scrubber. Significant depressions in yield occurred in the plants in the polluted greenhouse compared with those grown in purified air. Importantly, there were no observable lesions on the leaves, something that had previously been considered an indicator of damage [15].

Metal pollution was a further concern, with hotspots of metal deposition recorded close to smelters in Canada, the USA and at several locations in Europe [16–18]. Effects of metal pollution on vegetation were relatively local, within 30 km of the source, but the scale of emissions and persistent nature of some of the metal species left long-term soil contamination in these areas and damage to some flora and fauna. The research was largely concerned with the possibility of metals entering the human body via the ingestion of contaminated crops. Of wider concern was the use of tetraethyl lead in petrol, which had been introduced as an anti-knocking agent in many countries in the 1920s. By the middle of the twentieth century, it became apparent that lead was deposited alongside roads, with levels falling exponentially with distance from the road, but detectable up to 100 m away [19]. The identification of such widespread environmental contamination by lead resulted in massive pressure to reduce the use of tetraethyl lead, leading to a ban throughout much of the world.

Ground-level ozone became an important part of the global pollution climate in the mid-twentieth century. Trends in surface ozone from measurements over the last six decades [20]

show large, long-term trends in Europe. In presenting trends at remote sites, the effects of local sources are minimized and the regional trend becomes clearer. In the analysis by Cooper *et al.* [20], annual average surface ozone concentrations increased from 20 to 40 ppb between 1950 and 2000 and have changed little since then in Europe. Remote sites in North America and East Asia show increases of typically 10 ppb between 1980 and 2010, whereas the monitoring stations in the remote Pacific, Antarctica and Tasmania show quite small changes (less than 10 ppb) between 1980 and 2010 [20]. Ozone is produced within the troposphere through photochemical degradation of volatile organic compounds in the presence of NO_2 . Ozone became recognized as an important regional phytotoxic compound in the 1960s, initially in California [21], but extending to much of the USA and Southern Canada and detected in Europe during the 1970s [22].

Air pollution issues of the mid-twentieth century gradually extended to ever-larger areas as global emissions of sulfur approached peaks (figure 1). In part, these changes were due to the number and scale of sources, but also to the increasing heights of emission, with tall stacks on industrial sources of 200 m or more and large buoyant plumes of pollutant gases effectively injected towards the top of the boundary layer, promoting wide dispersion and long-range transport. These tall stacks, which were deployed to reduce ground-level pollution, essentially turned a local pollution problem into a regional one. Ecological effects of long-range transport of sulfur and nitrogen compounds began to be recognized in countries remote from the sources of the primary pollutants, initially as effects of acid deposition in Scandinavia [23] and in Scottish lochs [24]. By the end of the twentieth century, various national air quality laws and programmes had been implemented in many developed countries around the world. These had the effect of reducing emissions and deposition of many air pollutants, though emissions and deposition still remained higher than would be expected in the absence of anthropogenic emissions across much of the world.

Considering this background, we discuss four main threats for the global pollution climate in the late twentieth century:

1. Gas-phase effects of the primary emissions of SO_2 and NO_2 reflecting their emissions in urban areas and industrial regions including substantial tracts of arable cropland in Europe, North America, East and South Asia.
2. Effects of wet and dry deposited acidity from sulfur and nitrogen compounds, notably in upland areas where inputs are dominated by wet deposition. We give particular attention to the poorly buffered geological regions throughout Northern Europe, Scandinavia, the northeastern states of the USA and parts of Eastern Canada.
3. Ecosystem eutrophication by nitrogen deposition in both oxidized and reduced forms over much of northern Europe and large parts of eastern North America.
4. Ground-level ozone present at phytotoxic concentrations over much of the arable cropland of Europe and North America, East and South Asia.

In addition to the four main threats to vegetation more localized effects of metals, especially heavy metals, close to smelters and other metal processing industries, are notable through the twentieth century, but severe cases of damage to vegetation were not regional in scale. These four main pollutant threats will be examined via a series of case studies of major pollution impacts.

2. Gas-phase effects of SO_2 , NO_2 and NH_3 : case study lichen decline

In Europe at the beginning of the twentieth century, industrialization and large populations of cities produced high concentrations of SO_2 and NO_2 and thus urban air quality was typically very poor. The peaks in emissions and exposure of vegetation were between 1960 and 1970 for SO_2 in Europe and North America, thus during the middle decades of the twentieth century extensive areas of cropland and semi-natural areas of Europe and North America were exposed to damaging concentrations of SO_2 [25]. As a consequence of air pollution legislation, concentrations

of SO₂ declined steadily in many developed countries from about 1970. By 2015, there were no significant areas of Europe or North America experiencing concentrations of SO₂ at levels damaging to plants from anthropogenic sources [26]. In East Asia peak emissions were later, but since 2012 emissions of SO₂ in China have declined by approximately 40%. Emissions of NO₂ peaked around 1990 in Europe, rather later than SO₂ due to the rapid growth in vehicle usage, as well as industry and the slower introduction of effective controls on vehicle NO₂ emissions. Thus, concentrations of NO₂ remained high in urban and many rural areas through to the early years of the twenty-first century in Europe and North America [26].

Ammonia emissions, mainly from agricultural activities, steadily increased in all industrial countries through the twentieth century [27]. Few countries have made significant reductions in emissions, the Netherlands and Denmark being notable exceptions, each of which have reduced emissions over the last two decades by approximately 50% [28,29]. The overall global emissions of NH₃ continue to increase and as emissions of NH₃ are closely coupled to ambient temperature, it is likely that changes in climate will further increase emissions [27].

These broad changes in the chemical climate of industrial countries have had widespread effects on crops and natural vegetation, only some of which have been documented in detail. Here, the effects on lichens, which are particularly sensitive to atmospheric composition, are used as an example of the changing chemical climate of industrial countries.

Lichens are rarely a dominant feature of terrestrial habitats yet they are a ubiquitous component of most habitats globally, from arctic to desert to tropical forest, and in both managed and natural ecosystems. Although dependent on the substratum for their physical attachment, lichens have no vascular root system: they receive nutrients and water directly from the atmosphere, making them highly susceptible to changing atmospheric conditions.

In 1866, Nylander [30] observed that 'lichen deserts' were created around Paris, often extending for considerable distances aligned to the direction of the prevailing wind. Similar patterns were observed around that time across Europe, and in the mid-twentieth century around New York City [31] as well as other parts of the USA [32–34]. A cause of these declines was hypothesized as concentrations of acidifying pollutants in urban air. However, there were no systematically measured air quality data to support the observations. In the UK, it took the dramatic loss of life associated with the London smog in the winter of 1952 (estimated 12000 deaths [35]) to establish a national air quality recording system across both urban and rural areas; by the 1960s this comprised c. 1300 stations [36]. In the late 1960s, Hawksworth & Rose [37] correlated the distribution of lichen species across the UK with recorded levels of SO₂ from this network to create a 10-point scale of estimated SO₂ pollution based on species' sensitivity to air pollution. Since lichens absorb pollutants directly from the air, shrubby or hanging filamentous lichens with a large surface area such as *Usnea* species were at the top of the list of sensitive species, while crustose species associated with naturally acidic habitats such as *Lecanora conizaeoides* and leprose species of *Lepraria*, at the bottom of the list, were tolerant and becoming widespread in urban areas [37].

As SO₂ pollution declined, reactive nitrogen became the dominant atmospheric pollutant (figure 1), both as NO_x from vehicular and industrial emissions and as NH₃ from agricultural processes. The effect on lichen communities was dramatic, as species tolerant of nitrogen (nitrophytes) began to appear, in urban and agricultural areas. Gaseous ammonia deposited onto wet surfaces acts as a base, removing hydrogen ions from water and producing ammonium and hydroxide ions. Thus, in areas receiving high ammonia deposition, formerly common species tolerant of acid (acidophytes) became rare [38]. In areas affected by ammonia deposition, including naturally acidic moorland and heathland vegetation dominated by terricolous species of lichen, *Cladonia* died back [39,40]. The effect of ammonia is to increase bark pH, and research in the UK showed a strong correlation between lichen communities and bark pH and ammonia concentrations [41–43]. Quantitative studies of epiphytes on a range of tree species in sites where ammonia was monitored showed that lichen communities varied with tree species as well as air quality, so that acidophytes survived longer on acid-barked trees such as oak and birch, while nitrogen-tolerant species of the *Xanthorion* alliance appeared earlier on trees with a higher bark

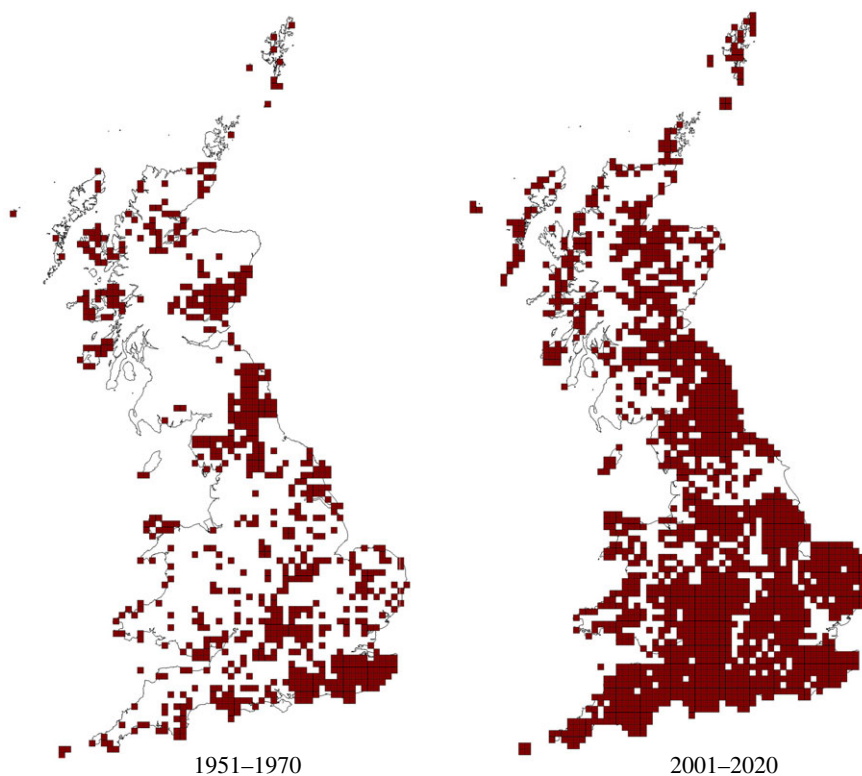


Figure 2. Distribution of the nitrogen-tolerant lichen *Xanthoria parietina* showing records in 10 km squares between 1951 and 1970 and between 2001 and 2020 (British Lichen Society). (Online version in colour.)

pH such as ash or poplar (figure 2) [41,43–47]. This shift in lichen communities from acidophyte species to nitrophyte species was demonstrated by a resurvey of twig communities on a nature reserve in Wales after 10 years which showed a loss of nitrogen-sensitive species on branches and the appearance of the *Xanthorion* [48]. A nationwide survey of sensitive and tolerant species of epiphytic macrolichens in the UK was used to test their response to measured ammonia at air pollution monitoring sites across the UK using lichen frequency and bark pH as measured variables. The results showed that a loss of nitrogen-sensitive species was taking place prior to the appearance of nitrophytes, and that this was happening at c. $1 \mu\text{g m}^{-3}$ NH_3 , well below the current critical level (concentration above which direct adverse effects are thought to occur) of $8 \mu\text{g m}^{-3}$ [49]. Since then $1 \mu\text{g m}^{-3}$ has been accepted as the critical level for lichens [50].

3. Deposited acidity in upland areas: case study forest decline

Prior to the late 1970s, concerns about acid deposition focused primarily on aquatic ecosystems. However, in the late 1970s and early 1980s, foresters and scientists began to note widespread diebacks of forest trees in both the northeastern USA and Europe, and suspicion mounted that acid deposition was the cause. The European decline was popularly known as ‘Waldsterben’ (forest death): it affected both hardwood and softwood trees and was characterized by extreme thinning of the crown, premature senescence, discoloration and loss of foliage, active casting of green foliage and loss of fine root biomass [51]. The North American decline was most severe in red spruce (*Picea rubens*) in mountainous locations of New York and New England [52]. Red spruce decline symptoms were the same on both continents and included reddening of needles in the early spring, a condition usually associated with winter freezing damage.

Most researchers on both continents pointed the finger at air pollution as the driving force behind the forest declines. Large concentrations of acidity in the orographic cloud and a windy upland climate for many forests leads to high deposition rates of the pollutants contained within the cloud water, which were shown to reduce the frost hardiness of red spruce [53]. Furthermore, Cape *et al.* [50] showed that it was the acidity and SO_4^{2-} ions rather than NO_3^- that were responsible for the observed reduction in frost hardiness. In Europe two major sets of hypotheses around the mechanisms gained prominence [51,54,55]. The ‘top-down’ hypotheses focused on the direct impacts to leaves and needles of high levels of pollutants, particularly oxides of sulfur and nitrogen, and ozone (e.g. [51,56]). Other pollutants such as metals and organic compounds were also implicated. The ‘bottom-up’ hypotheses proposed that the primary causal element of forest decline lies in the soil, through acidification and depletion of basic cations from soil exchange sites, leading to mobilization of toxic aluminium, as well as the excess enrichment by reactive nitrogen, leading to nutrient imbalances and acidifying nitrification pulses (e.g. [57,58]). Soluble aluminium causes a range of physiological damage, including direct mortality to fine roots and mycorrhizae, impaired ability to transport calcium and magnesium at the soil–root interface and a loss of the soil fauna that transport oxygen and nutrients to the deeper soil [59]. These two groups of hypotheses were generally not considered mutually exclusive, but which were the primary and which were the contributory factors was hotly debated. Since forest decline damage was ultimately recorded across a large number of hardwood and conifer species over a wide area and range of environmental conditions in Europe, it is highly likely that the primary and secondary drivers of damage also varied. However, to our knowledge, there has been no comprehensive analysis of the literature to provide an overview of the relative roles of these different mechanisms across European forests.

The cause of declines in red spruce in the northeastern USA was identified as foliar leaching of calcium, specifically membrane-bound calcium, causing reduced frost hardiness that could lead to winter damage [53,60,61]. The problem was particularly acute for red spruce because it occurred primarily in montane forests where it was exposed to cold winter temperatures and high deposition of acid rain and also to extremely acidic cloud droplets which are very effectively scavenged by conifer needles [62]. Low calcium availability in the soil could exacerbate the problem, as indicated by a study that showed that, compared to a reference catchment, frost damage was much reduced in a catchment in which calcium was added experimentally to the soil [63]. The same study showed a major improvement in the health of sugar maple (*Acer saccharum*) in the calcium-treated catchment, underscoring that it had also been affected by acid rain in the northeastern USA [64]. Researchers have since reported that many other species likely show similar effects, though possibly not as severe and not as widespread because they are less common across the landscape [65].

In recent decades, the decline in acid deposition in the eastern USA has led to a resurgence of red spruce, which is growing well throughout the region and expanding its range into lower-elevation forests [66,67]. Similarly, the International Co-operative Programme on the Assessment and Monitoring of Air Pollution Effects on Forests (ICP Forests) showed no major changes in the proportion of damaged trees in Europe overall from the mid-1990s to the mid-2010s, with conifers showing an improvement in condition and broadleaves showing a decline. Much of this was due to the new inclusion of Mediterranean ecosystems, where oak have shown a strong decline in recent years, mainly attributed to drought and insect damage [68]. In 2018 and 2019, a new decline in forest condition was reported in central and northern Europe; it is thought the primary driver is a widespread, persistent drought [69].

As sulfur emissions began to decline in North America and Europe, research focused increasingly on nitrogen deposition. The concept of ‘nitrogen saturation’ was advanced by Ågren & Bosatta [70] and Aber *et al.* [71] to indicate the deposition of nitrogen exceeding the biological capacity of the ecosystem to retain it. According to these studies, this leads to a sequence of changes in ecosystem function including increased plant tissue nitrogen concentrations, increased nitrogen cycling in the soil, increased nitrification, and ultimately elevated nitrogen leaching and soil acidification. Elevated nitrogen leaching was found to occur

in many catchments in the eastern USA where the deposition was above $8 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ [72] and in many forest plots in Europe where the deposition was above about $10 \text{ kg N ha}^{-1} \text{ yr}^{-1}$, especially if in the soil pH was low and % organic nitrogen was high [73]. The nitrogen saturation theory was further refined by Lovett & Goodale [74] who noted that the impacts of excess nitrogen deposition were not necessarily sequential, but rather all could occur simultaneously and their expression was controlled by the relative strength of the vegetation and soil sinks for nitrogen in the ecosystem. The nitrogen saturation theory marked a major shift in thinking about nitrogen by forest ecologists: previously nitrogen was primarily considered a key limiting nutrient, but the nitrogen saturation theory argued that when chronic deposition levels are high, it can also have adverse effects.

The impact of air pollution on forest health is now considered as a part of, or an acceleration to, a combination of stresses, especially drought [75,76]. Past management is also clearly a contributing factor: many parts of central Europe had a history of poor silviculture practices including even-aged, monocultural plantations with little genetic variability and litter-raking which removed valuable nutrients [75]. Thus the extreme acid deposition of the late 1970s may have been a 'final straw' for a large area of European forest already damaged and predisposed to further stress, triggering a widespread decline.

As pollutant emissions continue to decline, there may be a long-term legacy of damage done to soils by decades of acid deposition. Key studies show that forests where acidification is reduced can undergo major shifts in ecosystem pools and processes, including reductions in forest floor and upper mineral soil carbon and nitrogen pools and mobilization of nitrogen into deeper soil levels or export from the ecosystem [55,64,77,78]. The impact and duration of legacy effects in the ecosystem for both nitrogen and sulfur are determined by the extent to which the 'slow pools' in the ecosystem have been affected. These slow pools could include the available base cation pools that are replenished primarily by mineral weathering, or the stable soil organic matter pools that turn over very slowly [79,80]. As a result, we see some impacts of acid deposition reversing quickly (e.g. lower foliar leaching of nutrients, improved health of trees and declining stream nitrate fluxes and concentrations [81–83]) while other impacts, such as the depleted base saturation of acidified soils, may take decades or centuries to recover [84]. Thus, although the forests have appeared to stabilize and even improve in some areas, reducing acid deposition has also not removed all of the stresses.

4. Ecosystem eutrophication by nitrogen deposition: case study heathland decline and biodiversity losses

As described above, air pollution in Europe during the second half of the twentieth century is dominated by the gradual reduction in sulfur emissions and deposition and the slower decline in emissions and deposition of oxidized nitrogen. Notably, the deposition of nitrogen has been largely unchanged as emissions of NH_3 continued largely unabated [38]. The European hot spots for nitrogen deposition have been, and largely remain, the low countries of northern mainland Europe and the Netherlands in particular. Parts of the intensive livestock farming areas of the UK France and Denmark also receive high levels of nitrogen deposition and gas-phase NH_3 [42].

The most dramatic impacts of nitrogen deposition have been large community shifts in nutrient-poor terrestrial habitats such as grassland, heathland and bogs. In the Netherlands during the 1970s and 1980s, large areas of heathland transitioned to grassland. This typically occurred quite rapidly at a site (1–2 years as documented in one study [85]) and by 1983 80 km^2 of heathland in the Netherlands, nearly 20% of the total, had become dominated by grasses [86,87]. The grasses that became dominant were typically *Molinia caerulea* and *Deschampsia flexuosa*. It was observed that outbreaks of heather beetle (*Lochmaea suturalis*) caused serious damage to *Calluna vulgaris* which led to the death of heather plants and a replacement by grasses. This was first linked to eutrophication in 1977 by de Smit [88] and early experiments pointed

to the existence of relationships between eutrophication, heather beetle infestation and grass encroachment [86].

In the 1970s nitrogen deposition in the Netherlands ranged from around 40 to 80 kg N ha⁻¹ yr⁻¹ [89]. Experiments demonstrated, however, that even at N-addition levels much higher than ambient (200 kg N ha⁻¹ yr⁻¹), *M. caerulea* did not directly outcompete *C. vulgaris* [90]. Instead, fertilization by atmospheric deposition of N increased the quality of *C. vulgaris* as food plant for heather beetles. Heather beetle infestations on *C. vulgaris* cause many individual plants to die, which opens up the canopy and allows vigorously growing grasses to become dominant. During a heavy infestation of heather beetles little nitrogen is lost from the system: almost none is leached and only about 1 kg N ha⁻¹ is lost with dispersing insects [86]. Soils under dead *C. vulgaris* were found to have high mineralization and large N pools and low NH₄⁺ immobilization, and these conditions led to ammonium accumulation under dead heather, facilitating grass invasion [91].

Heather beetle larvae are able to detect and feed preferentially on the most N-rich shoots [92] and experiments showed that the beetles from fertilized plots showed higher growth rates compared to unfertilized plots and were the heaviest adults [85,93,94]. This was a consequence of better survival rates at larval stage and through hibernation induced by high nitrogen in the larval food supply [92]. The increased survival rates led to an increase in the severity (outbreak densities can be as high as 2000 beetles m⁻² [95]) and frequency of beetle outbreaks [93] with the frequency of outbreaks increasing from every 20 years to every 8 years [96]. It has since been shown that population numbers are highest after long-term nitrogen addition, indicating that impacts are likely to be exacerbated over time [92].

The conversion of heathland to grassland is very visible, but other impacts of nitrogen deposition on biodiversity are less apparent without surveys. One of the most reported of these is widespread losses of plant species richness. Declines in species richness with increasing nitrogen deposition had been observed in experimental situations for many years (e.g. [97,98]). In well-buffered prairie grasslands, long-term N-addition experiments in the field showed declines in species richness as a consequence of competition for other resources [99,100]. In 2004, it was demonstrated that the plant species richness of acidic grasslands was negatively correlated with ambient levels of nitrogen deposition across the UK. In acidic grasslands, the decline equated to one plant species lost for every additional 2.5 kg N ha⁻¹ yr⁻¹ [101] of long-term N deposition. Forbs in particular declined in species richness and cover, while cover of grasses tended to increase.

Since 2004 negative correlations between nitrogen deposition and plant species richness have been reported in a wide range of habitats (e.g. [102,103]) and in different parts of the world (e.g. [104–106]). These changes in species richness are accompanied by changes in the chemistry of soils, microbial communities (e.g. [107]) and soil processes (e.g. [108]). It is likely that changes in plant species richness and composition are driven by a combination of mechanisms including acidification, eutrophication and subsequent plant competition and interactions with secondary drivers. The relative importance of these drivers is likely influenced by habitat type, soil properties and climate as well as other variables. Such widespread changes in plant species richness are probably having impacts on invertebrate communities and higher up the food chain. There is growing evidence of these impacts [109] although this is an area where further research is needed. N deposition in some regions of the world continues to increase, but in parts of Europe, it is beginning to decline [26]. However, evidence from some long-term experiments indicates plant species richness and composition in many damaged areas are unlikely to recover without active management (e.g. [110–113]).

5. Ground-level ozone at phytotoxic concentrations

Ground-level ozone is formed in a series of photochemically driven reactions between oxides of nitrogen and volatile organic compounds. Unlike sources of SO₂ and NO₂, ozone is produced within the atmosphere, through the photochemical degradation of VOC in the presence of NO₂

[114]. The rate of production varies between the organic compounds as well as NO₂ mixing ratios and the VOC emitted by traffic provide some of the largest ozone production rates. Ozone production continues throughout the troposphere where the reactants and solar radiation are present and methane is a major contributor to tropospheric ozone production in remote regions [115]. Ozone is a hemispheric scale pollutant [114]: its effects on vegetation may occur much further from the sources of the precursor pollutants than the effects of SO₂ and NO₂. Ground-level ozone pollution is an issue in much of the mid-latitudes of the Northern Hemisphere as well as parts of the Southern Hemisphere, with the highest levels during summer in Southern Europe, South and East Asia and large areas of the southern states of the USA and Mexico [116]. Concentrations of ozone in the Northern Hemisphere mid-latitudes doubled between the late nineteenth century and 1980 and have continued to increase more slowly, reaching current levels of 35–40 ppb [116]. Ozone is a toxic pollutant to both agricultural crops and semi-natural vegetation. As a powerful oxidant, it exerts its effects primarily following stomatal uptake and the extent of injury is proportional to the absorbed stomatal dose [117]. Ozone impacts on crops are discussed in depth by Emberson [118].

Ozone impacts on vegetation were first recognized in Los Angeles, California in the 1950s. At the time poor air quality in the region was characterized by dense smogs which led to eye irritation and had a distinctive odour. These smogs were first linked to ozone by the chemist Haagen-Smit, who described the process of ozone formation by photochemical oxidation of hydrocarbons and nitrogen oxides [119]. The impacts of the smog on crops were already well described, but Haagen-Smit isolated ozone as the cause using laboratory fumigation experiments with spinach, beets, endive, oats and alfalfa. The results showed damage from ozone that was indistinguishable from that produced by the smog [119]. While the impacts of ozone on crops had already been described [120] this was the first time they were linked to smog.

Until the early 1970s, it had not been considered that ozone could ever become a problem in Britain due to the cooler climate, but then cases of extensive plant damage in the Rotterdam district of Holland occurred as a result of photochemical pollutants. Atkins *et al.* [22] subsequently detected summertime elevated ozone levels in a rural district of Berkshire UK, remote from any large conurbations. Bell & Cox [121] followed up these measurements by recording visible ozone damage on an ozone-sensitive cultivar of tobacco at the same location and showed that the injury correlated with ambient levels.

For the next two decades research into the impacts of ozone on plants focussed on damage to crops, with many species identified as showing visible damage and yield reductions [122]. Forest damage in the San Bernadino Mountains, California, in the early 1970s and visible signs of ozone stress in central European forests led to research into the impacts of ozone on young trees. This work led to clear evidence that ozone can affect the chlorophyll content, photosynthetic rate [123] and carbon allocation [124] of trees. Despite this, separating out the effects of ozone, acid deposition and other stressors on crown condition was a challenge. A number of regional- and continental-scale studies were devised to address this [125]. However, it was not until the 1990s that interest in the impacts of ozone widened from crops and trees. In 1998, a thorough review of ozone impacts on wild species highlighted the urgent need to investigate ozone impacts widely on semi-natural vegetation [125].

It remains the case that less is known about ozone impacts on semi-natural plants than agricultural crops, but understanding has developed considerably over the last two decades. Understanding the mechanisms of ozone impacts in semi-natural ecosystems is complicated by the need to account for species interactions which can lead to shifts in species composition and losses of biodiversity [126]. Wide variation in the responses of semi-natural species have been reported, as well as intra-specific variability in ozone sensitivity and heritable differences in ozone responses [126,127]. These heritable differences are likely a consequence of previous exposure [126]. Legumes [128] and summer annual plants [127] have been identified as particularly sensitive to ozone, but beyond this, there remains debate about whether sensitivity can be predicted from plant traits [126].

Predicting community responses remains a challenge. Mills *et al.* [129] used EUNIS (European Nature Information System) level 4 communities to predict community-wide ozone sensitivity from the published responses of individual species. They found all 54 of the EUNIS communities they considered had six or more sensitive species and were thus defined by the authors to be ozone sensitive. Grasslands had the most sensitive communities followed by heathland, scrub and tundra and mires, bogs and fens. However, Bassin *et al.* [130] caution that the risk to low-productivity perennial grasslands is commonly less than predicted from risk assessments based on individual species or immature mixtures in chambers.

6. Learning from the past and looking to the future

It is clear that we have come a long way in our understanding of the effects of atmospheric pollution on managed and natural vegetation. Pollution effects on plants have been observed for hundreds of years, but the identity of the pollutants responsible, the dose-response relationships, and causal mechanisms were unknown until the mid-twentieth century.

This paper has reviewed the evidence for four main threats from air pollution: (i) direct phytotoxic effects from gas-phase constituents, (ii) indirect effects from the deposition of acidifying agents, (iii) indirect effects from the deposition of nutrients and (iv) direct toxicity of ozone. The evidence suggests that each of these is not controlled by a single mechanism, nor do they operate in isolation. Rather, there are multiple pathways through which each mechanism may operate, many of which are occurring simultaneously, but to different degrees, within the same ecosystem [131,132]. For SO₂, direct phytotoxic effects appear to mostly be localized phenomena around point sources, although historically, when emissions were less regulated, direct effects were likely more widespread [11,12]. Acidifying and eutrophication effects appear to be widespread and simultaneous phenomena, but differ in relative magnitude based on a host of factors including the composition of the atmosphere. Ozone interactions are commonplace because ozone usually occurs in concert with other air pollutants such as sulfur and nitrogen oxides, which may affect plants simultaneously.

Although we have learned much about the possible outcomes related to increasing pollution deposition, we are only beginning to understand the reversibility of effects [79,113]. Pollution deposition has been decreasing through much of Europe and North America since the 1980s and 1990s and much more recently in parts of Asia [133]. These patterns provide an opportunity for 'natural experiments' to learn about the reversibility of these effects, in addition to the few controlled experiments. Early signs indicate that 'fast cycling' processes may recover fairly quickly (e.g. foliar nutrient balances, nitrate leaching, etc.), while 'slow cycling' processes may take several decades or more to recover (e.g. soil N pools, base cation availability, plant communities) [55,64,77,78]. In some cases, an alternative stable state may be reached and recovery may not occur over timescales relevant to public decision-making (e.g. several decades) without management intervention [79,134].

Looking forward, it is clear that there are multiple paths that different countries may follow either intentionally or unintentionally. Many developed nations are beginning to reduce pollution emissions and deposition through successful environmental policies, although the ultimate outcomes of many of those national policies on the environment are undetermined because the timescales of some anticipated changes are expected to take many decades and so far, insufficient time has passed. Developing nations may tread the same paths as developed nations, increasing both economic activity and air pollution emissions leading to regional reductions in biodiversity, soil acidification and eutrophication, only later to consider recovery, or they may evade the worst by learning from history. The axiom that 'an ounce of prevention is worth a pound of cure' is especially germane, as these developing nations, with the right international support and domestic incentives, have the opportunity to avoid many of these adverse effects of air pollution; for example, the costs of renewable energy have reached grid parity with many fossil fuels in many nations [135]. Regardless, the global scientific community has made significant strides in understanding and addressing the ecological effects from air pollution, though much work

remains as we assess the reversibility of these effects and the transferability of these lessons to understudied ecosystems.

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